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June 23rd, 2018 - The mole concept is one of the topics with which you kick off your preparation of physical chemistry in class 11 One of the reasons for it being taught earlier is that the concept of mole will be required in almost every other topic of physical chemistry that you study later irrespective of the complexity of that topic'

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June 21st, 2018 - One gram atom of any element contains the same number of atoms and one gram molecule of any substance contains the same number of molecules The value was found to be 6.022137×10^{23} The value generally used is 6.022×10^{23} '

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'Mole Concept Part 1 Class 11 Chemistry NCERT Unacademy

June 10th, 2018 - This lesson is about Mole concept and very helpful to study this confusing topic in simple way'

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July 13th, 2017 - just like you are taught counting in class first and adding up kg in kg for weight and liter in liter for liquids same way the number of atoms molecules are so large n number because of their very small size so to signify the count and make sense o" What is Mole Concept 2287674

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